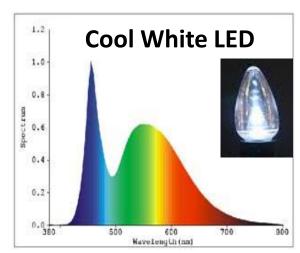
# Summary: how to graph light?

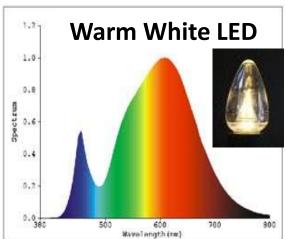
- "What color?" The apparent color of light is determined by the wavelength(s) of light waves.
- "How much?" The <u>intensity</u> of light is the amount of light energy falling on a surface per a unit of time.





The <u>spectrum</u> (<u>spectral composition</u>) of light is the relative light intensity for each wavelength present.





# **How to Make Light?**







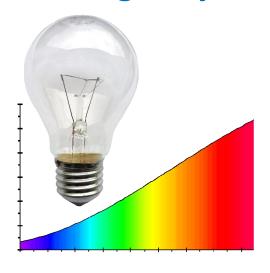


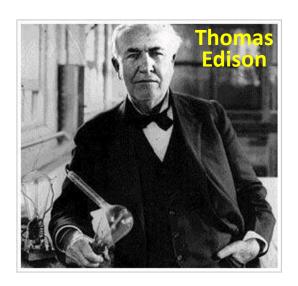
### Incandescence

Incandescence (from Latin "glowing white") is a special case of thermal radiation, specifically emission of visible light by a hot body.

Sunlight is the incandescence of the "white hot" surface of the Sun.







#### **Incandescent bulb:**

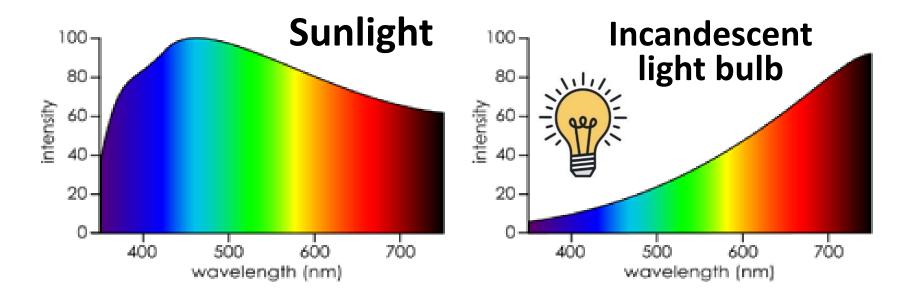
- electricity passes through a thin piece of metal wire called a filament
- the filament heats up and gives off thermal radiation composed of ~5% visible light and ~95% infrared light...
- ...very low energy efficiency!



## Incandescent Spectrum

"How much of each color is made?"

- X-AXIS: wavelength
- Y-AXIS: relative light intensity



Incandescent light sources produce light waves in a wide continuous range of wavelengths with gradually changing intensities; the spectrum is smooth.

# **How to Make Light?**











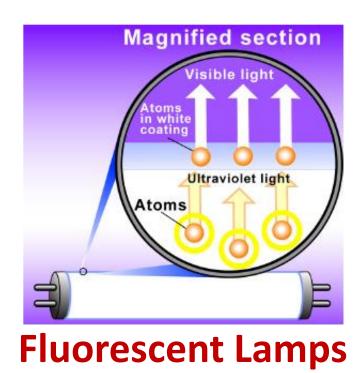
### Luminescence

<u>Luminescence</u> is emission of light by a substance not resulting from heat:

- Chemiluminescence (including bioluminescence), a result of a chemical reaction.
- Electroluminescence, emission of light due to electric current passed through a substance.



- Photoluminescence (fluorescence and phosphorescence) due to absorption of light with subsequent re-emission.
- Some other types.

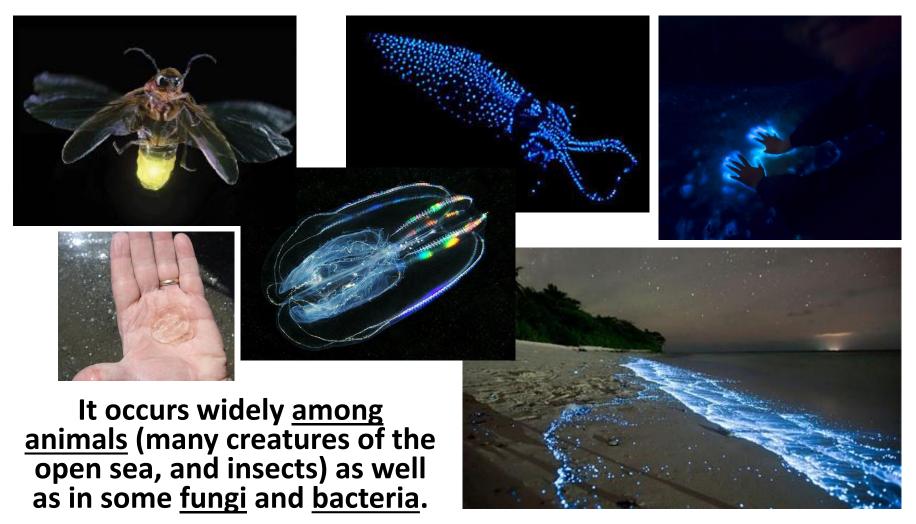


Glow

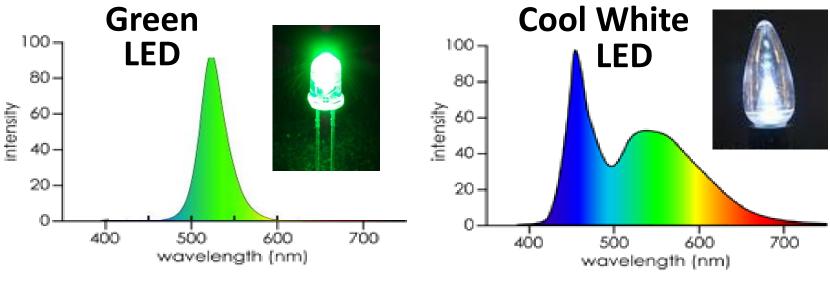
**Sticks** 

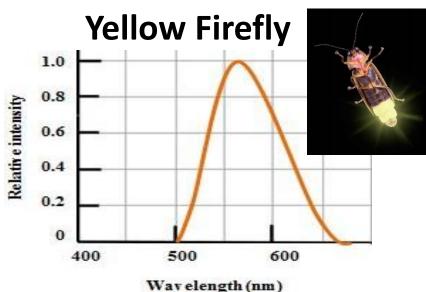
### **Bioluminescence**

<u>Bioluminescence</u> is <u>emission</u> of <u>light</u> by a <u>living</u> organism by means of a chemical reaction (type of *Chemiluminescence*).



# **Luminescent Spectrum**





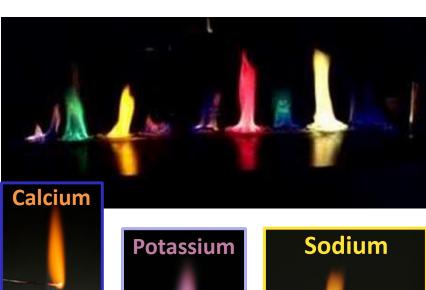
Luminescent light sources produce light waves in rather narrow "peaks" of wavelengths; the resulting light often appears to have a "distinct color".

#### Flame Test

- an analytic procedure used in chemistry to detect the presence of certain elements, primarily metal ions.







#### The idea:

- introduce a sample into flame to heat
- sample atoms sublimate (get isolated)
- since they are hot, they emit light
- specific colors are observed...



WHY?

