

3.6 HW Name \_\_\_\_\_

Directions: For each of the following compounds draw the Lewis Dot Diagram. - Carbon ALWAYS goes in the center.

$\text{NI}_3$	$\text{CF}_4$
$\text{HBr}$	$\text{SiBr}_4$
$\text{NH}_3$	$\text{F}_2$
$\text{Br}_2$	$\text{CH}_3\text{Br}$

$\text{CCl}_4$	$\text{PF}_3$
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$\text{H}_2\text{O}$	$\text{CF}_2\text{I}_2$
$\text{NCl}_3$	$\text{CHCl}_3$
$\text{CF}_2\text{H}_2$	$\text{OF}_2$

$\text{CO}_2$	$\text{O}_2$
$\text{N}_2$	$\text{CS}_2$